

Investigation of structure-properties relationship in a novel family of halogenoantimonates(III) and halogenobismuthates(III) with morpholinium cation:  $[\text{NH}_2(\text{C}_2\text{H}_4)_2\text{O}]\text{MX}_4$ . Crystal structure, phase transitions and dynamics of molecules.

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Three new organic-inorganic hybrids based on halogenoantimonates(III) and halogenobismuthates(III) with the morpholinium cation,  $[\text{NH}_2(\text{C}_2\text{H}_4)_2\text{O}]\text{SbCl}_4$ ,  $[\text{NH}_2(\text{C}_2\text{H}_4)_2\text{O}]\text{SbBr}_4$  and  $[\text{NH}_2(\text{C}_2\text{H}_4)_2\text{O}]\text{BiBr}_4$ , have been prepared and characterized with DSC, TGA, DTA and single-crystal X-ray diffraction. The common feature of the crystal structures of the studied compounds is the presence of polyanionic ( $[\text{MX}_4]^\infty(-)$ ) and morpholinium (head-to-tail configuration) chains, which expand themselves parallel to each other. The antimonate derivatives are isomorphous, crystallizing in a centrosymmetric orthorhombic *Pbca* space group and show no phase transitions (PTs) between 110 and 370 K. On the other hand,  $[\text{NH}_2(\text{C}_2\text{H}_4)_2\text{O}]\text{BiBr}_4$  undergoes two first-order structural PTs: I  $\leftrightarrow$  II at 321/343 K (cooling/heating) and II  $\leftrightarrow$  III at 285/289 K (cooling/heating). The mechanism of the PTs is discussed on the basis of crystallographic data and  $(1)\text{H}$  NMR and infrared spectroscopy. The PT at 343 K is accompanied by a spectacular switching of the spin-lattice  $T_1$  relaxation pathway. Structural parameters analysis has been performed to discuss a structure-properties relationship.

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