

Lanthanide complexes with phosphorylated 2-naphthylsulfonamides ligands as electromagnetic radiation converters.

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Streszczenie

The lanthanide complexes of the type $\text{Na}[\text{Ln}(\text{L}_4)]$ (**LnL1**) ($\text{Ln}=\text{Nd}^{3+}$, Eu^{3+} , Tb^{3+} , Yb^{3+}), with di(4-methylphenyl) (2-naphthyl)sulfonylamidophosphate (**HL¹**), as well as sodium salt (**NaL**), were synthesized. Single-crystal X-ray diffraction, IR, absorption and emission spectroscopies at 300, 77 and 4K were used to characterize the compounds. The ligand-to-metal energy transfer processes were analyzed. The dominant role of the ligand singlet state in intramolecular energy transfer processes was discussed and the main channels of the ligand-to-metal energy transfer for **EuL1** are proposed as $S_1 \rightarrow {}^5G_3$, 5G_6 , 5D_4 , 5L_6 with a fast contribution and $S_1 \rightarrow T_1 \rightarrow {}^5D_1$ with a slow contribution. The calculations of nonradiative rates of energy transfer covered the higher lying excited levels of Eu^{3+} (5D_J , 5L_J , 5G_J).

The participation of both the multipolar and exchange mechanisms was considered. The number of parameters determined for **LnL1** such as: experimental intensity parameters (Ω_λ), radiative (A_{rad}) and non-radiative (A_{nrad}) decay rates, emission lifetimes (τ), theoretical and experimental quantum yields were compared with the data for lanthanide complexes with dimethyl (2-naphthyl)sulfonylamidophosphate (**LnL2**). The strong temperature dependence of the excited 5D_4 level lifetime of **TbL1** was recorded in the range of 30 μs – 2ms, in the temperature range 300–77K. This phenomenon could be used in chemical thermometry.

Słowa kluczowe

Lanthanide, Phosphorylated sulfonamides, energy transfer, spectroscopy, crystal structure, Judd–Ofelt theory

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