

H-Ge bond activation by tungsten carbonyls: an experimental and theoretical study.

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Streszczenie

The activation of the Ge–H bond and the formation of several hydride complexes, characterized by high-field resonances, have been detected during the ^1H NMR spectroscopy monitoring of the photochemical reaction of Et_3GeH and Et_2GeH_2 with $\text{W}(\text{CO})_6$ and the norbornadiene complex $[\text{W}(\text{CO})_4(\eta^4\text{-nbd})]$. The activation of the Ge–H bond of triethylgermane in the photochemical reactions of tungsten(0) complexes has been applied in the hydrogermylation of norbornadiene (nbd), which leads to the formation of *endo*-triethylgermylnorbornene as the major product. The complex $[\{\text{W}(\mu\text{-}\eta^2\text{-H-GeEt}_2)(\text{CO})_4\}_2]$ has been fully characterized by NMR spectroscopy and by a single-crystal X-ray diffraction study. Evidence for the hydride ligand of the $\text{W}(\mu\text{-}\eta^2\text{-H-GeEt}_2)$ group has been provided by ^1H NMR spectroscopy ($\delta = -9.02$, $^1J_{\text{H-W}} = 31$ Hz) and by DFT calculations. A DFT study of the structural properties and ^1H NMR chemical shifts of several possible intermediate σ and hydride complexes formed during the photochemical reaction of $\text{W}(\text{CO})_6$ and Et_2GeH_2 has been performed.

Słowa kluczowe

Tungsten, Germane, σ -Bond activation, Hydrogermylation, Photolysis, DFT calculation

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