

Lanthanide carbonates.

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Streszczenie

The crystal and molecular structures of the rare earth carbonates with the general formulae $[C(NH_2)_3[Ln(CO_3)_4(H_2O)] \cdot 2H_2O$ (where $Ln = Pr^{3+}, Nd^{3+}, Sm^{3+}, Eu^{3+}, Gd^{3+}, Tb^{3+}$) and $[C(NH_2)_3[Ln(CO_3)_4] \cdot 2H_2O$ (where $Ln = Y^{3+}, Dy^{3+}, Ho^{3+}, Er^{3+}, Tm^{3+}, Yb^{3+}, Lu^{3+}$) were determined. The crystals consist of monomeric $[Ln(CO_3)_4(H_2O)]^{5-}$ or $[Ln(CO_3)_4]^{5-}$ complex anions in which the carbonate ligands coordinate to the Ln^{3+} ion in a bidentate manner. The spectroscopic (UV/Vis/NIR and IR) properties of the crystalline lanthanide carbonates, as well as their aqueous solutions, were determined. Correlation between the spectroscopic and the structural data enabled us to conclude that the $[Ln(CO_3)_4(OH)]^{6-}$ and $[Ln(CO_3)_4]^{5-}$ species predominate in the light and heavy lanthanide solutions, respectively. The nature of the $Ln-O$ interaction was also discussed. The experimental data, as well as the theoretical calculations, indicated that the $Ln-O(CO_3^{2-})$ bond is more covalent than the $Ln-O(OH_2)$ bond. Moreover, the covalency degree is larger for the heavy lanthanide ions. Inspection of the NBO results revealed that the oxygen hybrids, with the approximate composition sp^4 , form strongly polarized bonds with the $6s^6p^5d^4$ hybrids of lutetium.

Słowa kluczowe

structure elucidation, UV-vis, NIR spectroscopy, IR spectroscopy, density functional calculations

Adres publiczny

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