

Trinuclear and hexanuclear lanthanide(III) complexes of the chiral 3+3 macrocycle: X-ray crystal structures and magnetic properties.

Autorzy

Tomasz Bereta
Abhishake Mondal
Katarzyna Ślepokura
Yan Peng
Annie K. Powell

Jerzy Lisowski

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A new triphenolic hexaaza chiral macrocyclic amine **L** forms trinuclear complexes **1–3** with rare earth metal lanthanide(III) ions ($\text{Ln} = \text{Dy}, \text{Eu}, \text{and Y}$) with the general formula $[\text{Ln}_3\text{L}(\mu_3\text{-OH})_2(\text{NO}_3)_4(\text{H}_2\text{O})_2] \cdot x\text{CH}_3\text{OH}$. The crystal structures of the nitrate derivatives of this type reveal the presence of a $\{\text{Ln}_3(\mu_3\text{-OH})_2\}$ core within the macrocycle. For the chloride derivative of dysprosium(III) **4**, a duplex of the trinuclear compound is formed to give the hexanuclear $[\text{Dy}_6\text{L}_2(\mu_3\text{-OH})_3(\mu_3\text{-O})(\mu_2\text{-Cl})_3\text{Cl}_4(\text{H}_2\text{O})_2]$ compound, in which two trinuclear macrocyclic units are linked by bridging chloride anions, supplemented by a hydrogen bond connecting the central oxo and hydroxo bridges as well as by weak interactions at the periphery of the macrocycle. The nuclear magnetic resonance spectra of these complexes reveal a dynamic behavior in solution related to exchange of axial ligands and hindered rotation of phenyl substituents. Magnetic studies of the nitrate (**1–3**) and chloride (**4**) dysprosium(III) complexes suggest the presence of weak ferromagnetic interactions between neighboring metal centers. The interaction is strongest for compound **1**, and for the related duplex compound **4**, it appears to be somewhat weaker. The ac susceptibility measurements for complexes **1** and **4** confirm their field-induced single-molecule magnet behavior with the following characteristics: $U_{\text{eff}} = 10.6 \text{ cm}^{-1}$ (15.2 K), $\tau_0 = 2.05 \times 10^{-4} \text{ s}$ under 2500 Oe dc fields for **1**; $U_{\text{eff}} = 7.9 \text{ cm}^{-1}$ (11.4 K), $\tau_0 = 1.68 \times 10^{-4} \text{ s}$ under a 3000 Oe dc field for **4**.

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<https://www.acs.org/content/acs/en.html>