

One-pot synthesis, crystal structure and theoretical calculations of a dinuclear Mn(III) complex with *in-situ* generated O,N,O- and O,N-donor dichelating hydrazone ligand.

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Streszczenie

A new dinuclear Mn(III) complex, $[\text{Mn}_2(\text{L})_2(\text{CH}_3\text{OH})_2]$ (**1**), was synthesized by one-pot reaction of 2-aminobenzoylhydrazide, salicylaldehyde and $\text{Mn}(\text{OAc})_2 \cdot 4\text{H}_2\text{O}$ in methanol ($\text{H}_3\text{L} = (E)\text{-N}'\text{-(2-hydroxybenzylidene)-2-((E)-(2-hydroxybenzylidene)amino)benzohydrazide}$). The obtained compound was characterized by elemental analysis, spectroscopic methods, thermogravimetric analysis (TGA) and single crystal X-ray studies. Structural studies indicated that a ditopic O,N,O- and O,N-dichelating hydrazone ligand is obtained by *in-situ* condensation of 2-aminobenzoylhydrazide with salicylaldehyde which simultaneously is coordinated to two manganese(III) ions as trianionic ligand, $(\text{L})^{3-}$. During the formation of complex, Mn(II) is oxidized to Mn(III) by aerial oxygen and the Mn(III) ions have distorted octahedral geometry. The spectroscopic and structural properties of complex **1** were further studied by DFT calculations. DFT calculations on complex **1** indicated that the experimental geometric parameters and spectroscopic data are in the reasonable range obtained by theoretical studies.

Słowa kluczowe

Dinuclear manganese complex, *In-situ* reaction, crystal structure, DFT calculation, spectroscopic studies

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