

Synthesis, structure, and redox properties of N-confused bis(porphyrinato nickel(II)) linked byoxylene.

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Streszczenie

In a reaction of 5,10,15,20-tetraaryl-2-aza-21-carbaporphyrinato nickel(II) **2** with α,α' -dibromo-*o*-xylene, three different complexes containing a xylene moiety were obtained in the presence of a proton scavenger. The products were characterized by mass spectrometry, UV-vis, NMR, and, in the case of the dimeric complex **3**, X-ray crystallographic analysis (space group P, $a = 16.455(3)$ Å, $b = 16.776(3)$ Å, $c = 18.400(4)$ Å, $\alpha = 77.43(3)$ degrees, $\beta = 75.31(3)$ degrees, $\gamma = 66.20(3)$ degrees, $V = 4457.1(19)$ Å³, $Z = 2$). The monomeric species, diamagnetic **4** and paramagnetic **5**, contain one and two bromoxylene residues, respectively, while in **3**, the xylene bridge links two macrocyclic subunits, involving their internal carbons (C21) coordinated to diamagnetic nickel(II). Cyclovoltammetric measurements for *o*-xylene-linked bis(carbaporphyrinoid) **3** indicate cooperative effects resulting from an interaction between the subunits despite the isolation of their aromatic π -bond systems. An EPR-controlled titration of **3** with bromine allows consecutive detection of the mono- and bis-oxidized species (3Br , 3Br_2). The spectral patterns and spin-Hamiltonian parameters indicate metal-centered oxidation in 3Br ($g_x = 2.358$, $g_y = 2.150$, $g_z = 2.062$, $A(\text{Br})_x = 15$, $A(\text{Br})_y = 35$, $A(\text{Br})_z = 129$ G) and interaction of two electron spins of nickel(III) ions in 3Br_2 ($g_x = 2.328$, $g_y = 2.195$, $g_z = 2.065$, $D = 0.0173$ cm⁻¹, $E = -0.0018$ cm⁻¹, $A(\text{Br})_z = 63$ G). A product of the chemical reduction of a protonated form of the dimer was also detected by means of EPR ($g_1 = 2.298$, $g_2 = 2.218$, $g_3 = 2.192$), although no interaction between the nickel(I) centers can be observed for the reduced species.

Adres publiczny

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<https://www.acs.org/content/acs/en.html>

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