

Magnetic properties and molecular structure of a binuclear alternative bridged Cu(II)Re(IV) complex containing a macrocyclic ligand.

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Rok wydania

2014

Czasopismo

Polyhedron

Numer woluminu

75

Strony

1-8

DOI

10.1016/j.poly.2014.02.045

Kolekcja

Naukowa

Język

Angielski

Typ publikacji

Artykuł

Streszczenie

Two novel macrocyclic compounds, the heterobimetallic complex $\{(\text{CuL}_\alpha)[\text{ReCl}_4(\text{ox})]\}_n$ (**1**) and the mononuclear complex $[\text{CuL}_\alpha]\cdot 2\text{ClO}_4$ (**2**) (where $\text{L}_\alpha = N$ -meso-5,12-Me₂-7,14-Et₂-[14]-4,11-dieneN₄), have been synthesized and their crystal structures were determined by the single-crystal X-ray diffraction technique. Complex **1** crystallizes in the monoclinic space group $P2_1/c$, whereas **2** crystallizes in the monoclinic space group $P2_1/n$. The $[\text{CuL}_\alpha]^{2+}$ macrocyclic cation in **1** is coordinated from above and below by $[\text{ReCl}_4(\text{ox})]^{2-}$ units through the chloro- and monodentate oxalato ligands, and this creates an alternating chloro-oxalato-bridged heterometallic $\text{Re}^{\text{IV}}\text{-Cu}^{\text{II}}$ one-dimensional zig-zag chain. Their magnetic measurements were carried out over the temperature range 1.8–300 K using a Quantum Design SQUID magnetometer (MPMSxL-5 type). Compound **1** behaves like a ferrimagnetic $\text{Cu}^{\text{II}}\text{-Re}^{\text{IV}}$ dimetallic chain with two intrachain antiferromagnetic coupling parameters and strong single-ion anisotropy, $D(\text{Re}) = 109 \text{ cm}^{-1}$. Compound **2** shows weakly interacting paramagnetic centers in the crystal lattice. The effects of hydrogen bonds mediating the magnetic exchange interactions on the spin density have been evidenced by DFT calculations.

Słowa kluczowe

building block, bimetallic complexes, alternating chain, Exchange interaction, ferrimagnetic properties

Adres publiczny

<http://dx.doi.org/10.1016/j.poly.2014.02.045>

Strona internetowa wydawcy

<http://www.elsevier.com>