

## Chemistry of palladium phosphinite ( $\text{PPh}_2(\text{OR})$ ) and phosphonite ( $\text{P}(\text{OPh})_2(\text{OH})$ ) complexes: catalytic activity in methoxycarbonylation and Heck coupling reactions.

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The new phosphinite and phosphonite complexes (**1–8**) are very efficient catalysts for the methoxycarbonylation of iodobenzene and Heck cross-coupling of bromobenzene with butyl acrylate. High catalytic activity of these complexes can be explained by their *in situ* transformations during the reaction, stimulated by the presence of water, acid (HCl) or base ( $\text{NEt}_3$ ). Hydrolysis of phosphinite palladium complexes of the form  $\text{trans-PdCl}_2[\text{PPh}_2(\text{OR})]_2$  ( $\text{R} = \text{C}_6\text{F}_5$  **2**,  $^t\text{Bu}$  **3**, or O-menthyl **4**) results in the formation of the dimeric complex  $[\mu\text{-ClPd}(\text{PPh}_2\text{OH})(\text{PPh}_2\text{O})]_2$  **5**, which is deprotonated by  $\text{NEt}_3$ , producing a polymeric complex of formula  $[\text{Pd}(\text{P}(\text{O})\text{PPh}_2)_2]_n$  **8**. The reverse reaction, protonolysis of **8** with HCl, leads back to **5** and the monomeric complex **5a**. The phosphinite complex  $\text{PdCl}_2[\text{PPh}_2(\text{OBu})]_2$  **1** with a more lipophilic ligand,  $\text{PPh}_2(\text{OBu})$ , does not undergo hydrolysis under the same conditions. In the reaction of  $\text{PdCl}_2(\text{cod})$  with  $\text{P}(\text{OPh})_2(\text{OH})$ , the new dimer  $[\mu\text{-ClPd}(\text{P}(\text{OPh})_2\text{OH})(\text{P}(\text{OPh})_2\text{O})]_2$  **6** was obtained, whereas reaction of  $\text{Pd}(\text{OAc})_2$  with  $\text{P}(\text{OPh})_2(\text{OH})$  leads to the polymeric complex  $[\text{Pd}[\text{P}(\text{O})(\text{OPh})_2]_2]_n$  **7**. Protonolysis of **7** with HCl results in the formation of **6**.

### Adres publiczny

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### Strona internetowa wydawcy

<https://www.rsc.org/>