

Optical absorption spectra of divalent neodymium (Nd^{2+}) in bromide and iodide hosts.

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Rok wydania

2018

Czasopismo

European Journal of
Inorganic Chemistry

Strony

1660-1669

DOI

10.1002/ejic.201701315

Kolekcja

Naukowa

Język

Angielski

Typ publikacji

Artykuł

Streszczenie

Optical Absorption Spectra of Divalent Neodymium (Nd^{2+}) in Bromide and Iodide Hosts The electronic spectra for Nd^{2+} ions in bromide and iodide hosts are reported for the first time. These spectra, together with the recently obtained spectra for the chloride matrix, constitute a substantial series of Nd^{2+} spectra in crystals. Importantly, these crystals differ in the factors important for the physical properties of the divalent lanthanide Ln^{2+} ions; that is, the ligand's chemical character and the symmetry of the crystal field. The available experimental data allow systematic investigations of the impact of these factors on the electronic spectra, and thus, spectroscopic properties of Nd^{2+} ions. These investigations are the main aim of this study. The results indicate that Optical Absorption Spectra of Divalent Neodymium (Nd^{2+}) in Bromide and Iodide Hosts due to the increasing covalence of the Nd^{2+} environment in the $\text{Cl}-\text{Br}-\text{I}$ series, the energy of the first $f-d$ transition shifts into the infrared region, from 6121 cm^{-1} for $\text{SrCl}_2/\text{Nd}^{2+}$ to ca. 5260 cm^{-1} for $\text{SrI}_2/\text{Nd}^{2+}$, whereas the strength of the crystal field decreases along this series. Calculations using the parametric Hamiltonian and superposition model are carried out to quantitatively interpret the measured spectra. The simulated spectra exhibit good agreement with the experimental spectra. Using the proposed approach, a meaningful interpretation and assignment of the bands observed in the spectra has been achieved.

Słowa kluczowe

neodymium, structure elucidation, ligand effects, lanthanides, divalent neodymium

Adres publiczny

<https://doi.org/10.1002/ejic.201701315>

Plik został wygenerowany dnia 2026-05-02 09:58:07

Adres w repozytorium <https://old.chem.uni.wroc.pl/pl/repozytorium/bz3PWQG>.