

## Gas phase FT-IR spectra and structure of aminoalcohols with intramolecular hydrogen bonds. I. The shape of the $\nu(\text{OH})$ vibrational bands in $\text{R}_2\text{NC}_3\text{H}_6\text{OH}$ ( $\text{R}=\text{H}, \text{CH}_3$ ).

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### Streszczenie

Gas phase FT-IR spectra of 3-dimethylamino-1-propanol (**1**) and 3-amino-1-propanol (**2**) are recorded. It is found that the forms with intramolecular hydrogen bonds strongly predominate up to 100°C in both compounds. The origin of multicomponent structure of  $\nu(\text{OH})$  bands is discussed on the assumption of conformational equilibria as well as the coupling between the  $\nu(\text{OH})$  vibration mode with the low frequency chelate ring deformation modes affecting the  $\text{N}\cdots\text{O}$  distance. DFT ab initio calculations are performed at 6-31G(d,p) and 6-311++G(2d,2p) levels. It is concluded that the  $\nu(\text{OH})$  band structure in **1** results from high/low frequency modes coupling.

### Słowa kluczowe

Ab initio, Aminopropanol, Conformation, Gas phase IR spectra, Hydrogen bond, Multicomponent bands structure

### Adres publiczny

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