

Structure and characterization of copper(II) perchlorate with diethyl(pyridin-2-ylmethyl) phosphonate (2-pmpe)ligand: $[\text{Cu}(2\text{-pmpe})_2(\text{ClO}_4)_2]$.

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Streszczenie

Copper(II) mononuclear compound $[\text{Cu}(2\text{-pmpe})_2(\text{ClO}_4)_2]$ (2-pmpe = diethyl(pyridin-2-ylmethyl) phosphonate) was synthesized and studied. Examination of the crystal structure by the X-ray diffraction method revealed two crystallographically unrelated molecules, $[\text{Cu}(1)(2\text{-pmpe})_2(\text{ClO}_4)_2]$ (1) and $[\text{Cu}(2)(2\text{-pmpe})_2(\text{ClO}_4)_2]$ (2) in an asymmetric part of the unit cell. The geometry about the Cu(1) and Cu(2) chromophores shows elongated octahedra, resulting from the didentate N,O-bonded two chelate 2-pmpe ligands and two coordinated perchlorate ions around the Cu(II) cations (CuN₂O₄ chromophore). Similarly to 1, molecules 2 are linked to each other by weak C–H···O hydrogen bonds and $\pi\cdots\pi$ stacking interactions. Additionally, both 1 and 2 molecules are linked to each other through weak C–H···O hydrogen bonds and C–H··· π contacts, resulting in a 3D polymeric network arrangement. Magnetic data indicate a very weak intermolecular exchange interaction between copper(II) ions ($zJ' = -0.20 \text{ cm}^{-1}$) transmitted through non-covalent interactions in the crystal lattice. The spectral properties are in accordance with the structural and magnetic data.

Słowa kluczowe

copper(II), phosphonic acid ester, crystal structure, spectroscopy, magnetism

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