

Quantum chemical topology description of the hydrogen transfer between the ethynyl radical and ammonia ($C_2H\cdot + NH_3$) - the electron localization function study.

Autorzy

Sławomir Berski

Zdzisław Latajka

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Streszczenie

The topological analysis of the Electron Localization Function (ELF) has been carried out for the hydrogen abstraction pathway

between ethynyl radical ($HCC\cdot$) and ammonia (NH_3). The IRC path, minima ($HCC-NH_2$, $HCCH-NH_2$) and transition structure

have been calculated at the UB3LYP/6-311++G(2d,2p) computational level. The 'closed-shell' and 'spin-polarized' formula of ELF are

used for an analysis. A detailed study of the topology of ELF (the basin population and spin density redistribution) shows that the reaction consists of three main steps distinguished on the IRC path. First, the N-H bond in ammonia is broken, then the hydrogen atom with a population of ca. 0.7 e is transferred and finally the C-H bond in acetylene is formed. It is interesting to note that all chemically important changes occur after the transition state.

Adres publiczny

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