

Impact of aluminium ions on adriamycin-type ligands.

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Potentiometric and spectroscopic measurements have shown that Al^{3+} ions form 1:1 complexes with adriamycin and its analogues. The major complex at $\text{pH} > 5$ is a tetrahedral species in which the metal ion is co-ordinated to the ligand *via* two anthracycline oxygens. Two OH^- groups complete the co-ordination sphere. Although the complexes are comparatively stable the formation of $[\text{Al}(\text{OH})_4]^-$ at $\text{pH} > 8.5$ excludes adriamycin from the metal ion co-ordination sites. The unusual fluorometric behaviour of the drug molecule is also present in the complex formed at lower pH.

Adres publiczny

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