

A new family of chiral chelating diamines with transition-metal stereocenters: synthesis, structure, and reactivity of the enantiomerically pure dirhenium-substituted 1,2-diamine $(\eta^5\text{-C}_5\text{H}_5)\text{Re}(\text{NO})\text{-(PPh}_3\text{)(CH}_2\text{N(CH}_3\text{)CH}_2\text{CH}_2\text{(H}_3\text{C)NCH}_2\text{)(Ph}_3\text{P)(ON)Re}(\eta^5\text{-C}_5\text{H}_5\text{)}$.

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Streszczenie

Reaction of enantiopure $(S)\text{-(}\eta^5\text{-C}_5\text{H}_5\text{)Re(NO)(PPh}_3\text{)(CH}_3\text{)}$ ($(S)\text{-1}$) with $\text{Ph}_3\text{C}^+\text{BF}_4^-$ (1 equiv) and then the N,N' -dimethyl 1,2-diamine $\text{HN(CH}_3\text{)CH}_2\text{CH}_2\text{(H}_3\text{C)NH}$ (0.5 equiv) gives the diammonium salt $(S_{\text{Re}}S_{\text{Re}})\text{-}[(\eta^5\text{-C}_5\text{H}_5)\text{Re(NO)(PPh}_3\text{)(CH}_2\text{NH(CH}_3\text{)CH}_2\text{CH}_2\text{(H}_3\text{C)HNCH}_2\text{)(Ph}_3\text{P)(ON)Re}(\eta^5\text{-C}_5\text{H}_5)]^{2+}(\text{BF}_4^-)_2$ (94%) as a mixture of Re/N configurational diastereomers. Reaction with $t\text{-BuOK}$ yields the title compound $(S_{\text{Re}}S_{\text{Re}})\text{-4}$ (66%) as an air-stable orange powder. Reaction with $(\text{PhCN})_2\text{PdCl}_2$ gives a single diastereomer of a chelate complex, (80%), the configuration and approximate C_2 symmetry of which has been established crystallographically. Racemic **1**, $\text{Ph}_3\text{C}^+\text{BF}_4^-$ (1 equiv), and the N,N' -dimethyl 1,2-diamine $\text{HN(CH}_2\text{CH}_3\text{)CH}_2\text{CH}_2\text{(H}_3\text{CCH}_2\text{)NH}$ (1 equiv) give $[(\eta^5\text{-C}_5\text{H}_5)\text{Re(NO)(PPh}_3\text{)(CH}_2\text{NH(CH}_2\text{CH}_3\text{)CH}_2\text{CH}_2\text{NH(CH}_2\text{CH}_3\text{))}]^+\text{BF}_4^-$, and the crystal structure of the $S_{\text{Re}}S_{\text{N}},R_{\text{Re}}R_{\text{N}}$ diastereomer is determined. The ReCH_2N conformations of the preceding compounds, and their influence upon the diastereoselectivities, are analyzed in detail

Słowa kluczowe

Cations, Coordination compounds, Molecular structure, Nitrogen, rhenium

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<https://www.acs.org/content/acs/en.html>

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