

Nickel complexes of 21-oxaporphyrin and 21,23-dioxaporphyrin.

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Streszczenie

The nickel(I) and nickel(II) complexes of 5,20-bis(p-tolyl)-10,15-diphenyl-21-oxaporphyrin (ODTDPPH) and 5,10,15,20-tetraphenyl-21,23-dioxaporphyrin (O2 TPP) have been investigated. These oxa analogues of 5,10,15,20-tetraarylporphyrin, where one or two pyrrole rings are replaced by a furan moiety, have been synthesized by condensation of the respective precursors, namely 2,5-bis(arylhydroxymethyl)furan, pyrrole, and arylaldehyde. Insertion of nickel(II) into ODTDPPH or O2 TPP yielded high-spin five- and six-coordinate $[(\text{ODTDPP})\text{Ni}(\text{II})\text{Cl}]$ and $[(\text{O2 TPP})\text{Ni}(\text{II})\text{Cl}_2]$ complexes, which can be reduced with moderate reducing reagents. The EPR spectra of $[(\text{ODTDPP})\text{Ni}(\text{I})]$ and $[(\text{O2 TPP})\text{Ni}(\text{I})\text{Cl}]$ revealed the Ni(I) oxa(dioxa)porphyrin rather than a Ni(I) anion radical electronic structure. In the structures of $[(\text{ODTDPP})\text{Ni}(\text{II})\text{Cl}]$, $[(\text{O2 TPP})\text{Ni}(\text{II})\text{Cl}_2]$, and $[(\text{ODTDPP})\text{Ni}(\text{I})]$, determined by X-ray diffraction, the furan ring is planar and coordinates in the $\eta(1)$ fashion through the trigonal oxygen atom; the nickel ion lies in the furan plane for the latter two complexes, but slightly outside it in $[(\text{ODTDPP})\text{Ni}(\text{II})\text{Cl}]$. The Ni-N and Ni-O bond lengths decrease upon reduction of high-spin five-coordinate $[(\text{ODTDPP})\text{Ni}(\text{II})\text{Cl}]$ to four-coordinate $[(\text{ODTDPP})\text{Ni}(\text{I})]$. The pattern of downfield pyrrole resonances in (1) H NMR spectra of $[(\text{ODTDPP})\text{Ni}(\text{II})\text{Cl}]$ and $[(\text{O2 TPP})\text{Ni}(\text{II})\text{Cl}_2]$ has been established. The downfield positions of furan resonances are unusual for Ni(II) heteroporphyrins; they have been accounted for by the nearly in-plane coordination of the furan moiety as opposed to the side-on coordination found for thiophene- or selenophene-containing heteroporphyrins. An example of ion-pair formation, $[(\text{O2 TPPH})_2][\text{Ni}(\text{II})\text{Cl}_4]$, was produced from $[(\text{O2 TPP})\text{Ni}(\text{II})\text{Cl}_2]$ by acidification with HCl.

Słowa kluczowe

heterocycles, heteroporphyrins, nickel, porphyrinoid

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