

## Homogeneous/heterogeneous palladium based catalytic system for Heck reaction. The reversible transfer of palladium between solution and support.

### Autorzy

Iweta Pryjomska-Ray

Andrzej Gniewek

Anna M. Trzeciak

Józef J. Ziółkowski

Włodzimierz Tylus

### Rok wydania

2006

### Czasopismo

Topics in Catalysis

### Numer woluminu

40

### Strony

173-184

### DOI

10.1007/s11244-006-0119-1

### Kolekcja

Naukowa

### Język

Angielski

### Typ publikacji

Artykuł

### Streszczenie

Pd(II)/Al<sub>2</sub>O<sub>3</sub> and Pd(0)/Al<sub>2</sub>O<sub>3</sub>, containing 0.6% of palladium were characterized by EDX, SEM and XRD methods and used as catalysts of the Heck coupling of bromobenzene with butyl acrylate at 140°C in molten [Bu<sub>4</sub>N]Br salt. Monoarylated (trans-PhCH = CHC(O)OBu) (1) and diarylated (Ph<sub>2</sub>C = CHC(O)OBu) (2) products were obtained in amounts dependent on kind of base present in the system (Scheme 1). During the reaction palladium was partially leached from the support forming [Bu<sub>4</sub>N][PdBr<sub>4</sub>] complex that catalyze Heck reaction or undergo readsorption on Al<sub>2</sub>O<sub>3</sub>. These soluble palladium complexes are partially reduced to Pd soluble nanoparticles which can also be anchored on Al<sub>2</sub>O<sub>3</sub> giving active catalyst of Pd(0)/Al<sub>2</sub>O<sub>3</sub> type. Reduction of Pd(II) to Pd(0) during catalytic process at the presence of bases (NaHCO<sub>2</sub>, NaHCO<sub>3</sub>, NaOAc, Cs<sub>2</sub>CO<sub>3</sub>) was studied by XPS method and the total reduction was observed in reactions of PdCl<sub>2</sub>(PhCN)<sub>2</sub> with NaHCO<sub>2</sub> or with NaHCO<sub>3</sub> and [Bu<sub>4</sub>N]Br. It was experimentally proved that heterogenized catalyst, Pd(0)/Al<sub>2</sub>O<sub>3</sub>, after oxidative addition of aryl halides, serve as a source of soluble palladium species and colloidal nanoparticles that are active as homogeneous catalysts.

### Słowa kluczowe

Heck reaction, XPS, XRD, SEM, heterogenized catalysts, Pd colloid, Pd nanoparticles.

### Adres publiczny

<http://dx.doi.org/10.1007/s11244-006-0119-1>

### Strona internetowa wydawcy

<http://link.springer.com>