

## Impact of $\alpha$ -hydroxymethylserine (HmS) residue on the binding ability of histydył residue in the HmS-His dipeptide towards $\text{Cu}^{\text{II}}$ , $\text{Ni}^{\text{II}}$ and $\text{Zn}^{\text{II}}$ .

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Potentiometric and spectroscopic data have shown that  $\alpha$ -hydroxymethylserylhistidine is a very efficient ligand for  $\text{Cu}^{\text{II}}$ ,  $\text{Ni}^{\text{II}}$  and  $\text{Zn}^{\text{II}}$ . The stabilities of the complexes formed are considerably higher than those obtained for Gly-His or Ala-His dipeptides. Copper(II) and  $\text{Ni}^{\text{II}}$  form very stable tetrameric complexes  $\text{M}_4\text{H}_{-8}\text{L}_4$ . According to  $^1\text{H}$  NMR spectra the nickel tetrameric complex is of  $\text{C}_2$  symmetry with two pairs of different imidazole rings. Zinc(II), on the other hand, forms only monomeric species but due to the second His residue it forms very stable  $\text{ZnH}_{-1}\text{L}$  species *via* a  $\{\text{NH}_2, \text{N}^-_{\text{amide}}, \text{N}_{\text{imidazole}}\}$  donor set.

### Adres publiczny

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