

Structural, spectroscopic and catalytic properties of water-soluble hydride rhodium complexes $[\text{RhH}(\text{Rtpa}^+ \text{I}^-)_4]\text{H}_2\text{O}$ (R = Me, Et).

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New rhodium(I) hydrido complexes $[\text{RhH}(\text{mtpa}^+\text{I}^-)_4]\cdot\text{H}_2\text{O}$ (**1**) and $[\text{RhH}(\text{etpa}^+\text{I}^-)_4]\cdot\text{H}_2\text{O}$ (**2**) (where mtpa^+I^- =1-methyl-1-azonia-3,5-diaza-7-phosphatricyclo[3.3.1.1^{3,7}]decane iodide, etpa^+I^- =1-ethyl-1-azonia-3,5-diaza-7-phosphatricyclo[3.3.1.1^{3,7}]decane iodide) have been synthesized and characterized using ^1H , ^{31}P NMR and IR spectra and X-ray crystallography. Both complexes have the trigonal-bipyramidal structure. The ^1H NMR spectra indicate that in solutions the complexes have fluxional properties. The compounds are very stable in water under neutral atmosphere. They are efficient catalysts of hydrogenation of C=C bonds in two-phase catalytic systems

Słowa kluczowe

Crystal structures, Alkene hydrogenation, Rhodium complexes, Hydride complexes, Water-soluble complexes

Adres publiczny

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