

Infrared spectroscopy and ab initio study of hydrogen bonded $\text{Cl}_3\text{CD}\cdot\text{N}(\text{CH}_3)_3$ complex in the gas phase.

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Streszczenie

FTIR spectra of the gas phase $\text{Cl}_3\text{CD}+\text{TMA}$ mixture have been studied at room temperature in $\sim 800\text{-}4000\text{ cm}^{-1}$ frequency domain. The formation of the H-bonded $\text{Cl}_3\text{CD}\dots\text{TMA}$ complex has been detected. Spectroscopic parameters of the band ascribed to the complex were evaluated. MP2 frozen core ab initio calculations have been carried out with the Pople-type 6-311++G(d,p) basis set. The equilibrium geometries and harmonic vibrational frequencies of the complex were obtained using CP-corrected gradient techniques. The "freq=anharm" option has been tested for Cl_3CD monomer and $\text{Cl}_3\text{CD}\dots\text{TMA}$ complex to examine possible anharmonic effects on the vibrations localized on the proton donor. The effects of Darling-Dennison and Fermi resonances on the frequency of the stretching vibration of the CH proton donor were analyzed.

Słowa kluczowe

FTIR spectroscopy, Gas phase, Chloroform, H-bond, Fermi resonance, Anharmonicity

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