

Oxidation and oxygenation of iron complexes of 2-aza-21-carbaporphyrin.

Autorzy

Krystyna Rachlewicz
S.-L. Wang
J.- L. Ko
Chen-Hsiung Hung
Lechosław Latos-Grażyński

Rok wydania

2004

Czasopismo

Journal of the American
Chemical Society

Numer woluminu

126

Strony

4420-4431

DOI

10.1021/ja039792y

Kolekcja

Naukowa

Język

Angielski

Typ publikacji

Artykuł

Streszczenie

Oxidation and oxygenation of $(HCTPPH)Fe^{II}Br$ an iron(II) complex of 2-aza-5,10,15,20-tetraphenyl-21-carbaporphyrin $(CTPPH)_2$ have been followed by 1H and 2H NMR spectroscopy. Addition of I_2 or Br_2 to the solution of $(HCTPPH)Fe^{II}Br$ in the absence of dioxygen results in one-electron oxidation yielding $[(HCTPPH)Fe^{III}Br]^+$. One electron oxidation with dioxygen, accompanied by deprotonation of a C(21)H fragment and formation of an Fe–C(21) bond, produces an intermediate-spin, five-coordinate iron(III) complex $(HCTPP)Fe^{III}Br$. In the subsequent step an insertion of the oxygen atom into the preformed Fe^{III}–C(21) bond has been detected to produce $[(CTPPO)Fe^{III}Br]^-$. Protonation at the N(2) atom affords $(HCTPPO)Fe^{III}Br$. The considered mechanism of $(HCTPPH)Fe^{II}Br$ oxygenation involves the insertion of dioxygen into the Fe–C bond. The 1H NMR and 2H NMR spectra of paramagnetic iron(III) complexes were examined. Functional group assignments have been made with use of selective deuteration. The characteristic patterns of pyrrole and 2-NH resonances have been found diagnostic of the ground electronic state of iron and the donor nature localized at C(21) center as exemplified by the 1H NMR spectrum of intermediate-spin $(HCTPP)Fe^{III}Br$: β -H 7.2, –10.6, –19.2, –20.6, –23.2, –24.9, –43.2; 2-NH –76.6 (ppm, 298 K). The structures of two compounds $(HCTPP)Fe^{III}Br$ and $(HCTPPO)Fe^{III}Br$, were determined by X-ray diffraction studies. In the first case, the iron(III) is five-coordinate with bonds to three pyrrole nitrogen atoms (Fe–N distances: 1.985(8), 2.045(7), 2.023(8) Å), and the pyrrolic trigonal carbon (Fe–C: 1.981(8) Å). The iron(III) of $(HCTPPO)Fe^{III}Br$ forms bonds to three pyrrole nitrogen atoms (Fe–N distances 2.104(5), 2.046(5), 2.102(5) Å). The Fe–O 2.041(5) Å and Fe–C(21) 2.192(5) Å distances suggests a direct interaction between the iron center and the π electron density on the carbonyl group in a η^2 fashion.

Słowa kluczowe

Iron, Nuclear magnetic resonance spectroscopy, Oxidation,
Pyrroles, Saturation

Adres publiczny

<https://doi.org/10.1021/ja039792y>

Strona internetowa wydawcy

<https://www.acs.org/content/acs/en.html>

Plik został wygenerowany dnia 2026-05-04 18:06:19

Adres w repozytorium <https://old.chem.uni.wroc.pl/pl/repozytorium/KnMOzfw>.