

## Ammonium *tris* oxalatoferrate(III) as a source of metalloligand in direct synthesis of Cu/Fe coordination polymer.

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It has been shown that one-pot reaction of copper powder, ammonium tris-oxalatoferrate(III)  $(\text{NH}_4)_3[\text{Fe}(\text{C}_2\text{O}_4)_3]\cdot 3\text{H}_2\text{O}$  as a source of metalloligand, and ethylenediamine (en) leads to the formation of heterobimetallic complex  $(\text{NH}_4)[\text{Cu}(\text{en})_2\text{Fe}(\text{C}_2\text{O}_4)_3]\cdot 2\text{dmsO}$  (1). This complex has been characterised by elemental analysis, IR spectroscopy and single crystal X-ray crystallography. Crystal structure analysis reveals that 1 consists of  $[\text{Cu}(\text{en})_2\text{Fe}(\text{C}_2\text{O}_4)_3]^-$  anionic chains with regularly alternated  $[\text{Cu}(\text{en})_2]^{2+}$  and  $[\text{Fe}(\text{C}_2\text{O}_4)_3]^{3-}$  moieties. Magnetic susceptibility measurements indicate absence of a significant exchange interaction between the metal units. The polycrystalline X-band EPR spectrum of 1 exhibits broad line characteristic of Fe(III) centers, whereas the signals observed in EPR spectrum of frozen dmf solution of 1 are distinctly associated with zero field splitting in the spin states of rhombically distorted Fe(III) centers, non interacting with Cu(II) centers.

### Słowa kluczowe

Direct synthesis, heterobimetallic complexes, copper, tris-oxalatoferrate(III), X-ray crystallography, X-band EPR

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