

## Five-coordinate complexes of 21-thiaporphyrin : preparations, spectra, and structures of iron(II), nickel(II), and copper(II) complexes.

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Insertion of Fe(II), Ni(II), and Cu(II) as their chloride salts into tetraphenyl-21 -thiaporphyrin, STPPH, yields  $\text{Fe}(\text{STPP})\text{Cl}$  ( $S = 2$ ,  $M_{\text{eff}} = 5.2 \text{ mb}$ ),  $\text{Ni}(\text{STPP})\text{Cl}$  ( $S = 1$ ,  $M_{\text{eff}} = 3.3 \text{ mb}$ ) and  $\text{Cu}(\text{STPP})\text{Cl}$  ( $S = 1/2$ ). Electronic spectra of these complexes show porphyrin-like behavior with a strong Soret peak at 400 nm. The Mössbauer spectrum for  $\text{Fe}(\text{STPP})\text{Cl}$  ( $Q_S = 3.93 \text{ mm/s}$ ,

$IS = 0.88 \text{ mm/s}$  at 295 K) is characterized by a large quadrupolar splitting found for related high-spin iron(II) complexes. The

structures of these three isomorphous compounds (triclinic,  $P\bar{1}$ ) have been determined by X-ray diffraction:  $\text{Fe}(\text{STPP})\text{Cl}(\text{CH}_3\text{CN})$ ,

$a = 10.059 \text{ \AA}$ ,  $b = 13.661 (4) \text{ \AA}$ ,  $c = 15.033 (5) \text{ \AA}$ ,  $\alpha = 102.53 (2)^\circ$ ,  $\beta = 104.16 (2)^\circ$ ,  $\gamma = 106.37 (2)^\circ$ ,  $Z = 2$  at 130

K, least-squares refinement of 240 parameters using 3034 reflections,  $R = 0.077$ ;  $\text{Ni}(\text{STPP})\text{Cl}$ ,  $a = 10.154 (3) \text{ \AA}$ ,  $b = 13.687$

$(4) \text{ \AA}$ ,  $c = 15.064 (4) \text{ \AA}$ ,  $\alpha = 101.97 (2)^\circ$ ,  $\beta = 103.62 (2)^\circ$ ,  $\gamma = 105.90 (2)^\circ$ ,  $Z = 2$  at 293 K, least-squares refinement of 489 parameters with 4860 reflections,  $R = 0.053$ ;  $\text{Cu}(\text{STPP})\text{Cl}(\text{CH}_3\text{CN})$ ,  $a = 10.061 (6) \text{ \AA}$ ,  $b = 13.646 (7) \text{ \AA}$ ,  $c = 14.930 (8) \text{ \AA}$ ,

$\alpha = 102.12 (4)^\circ$ ,  $\beta = 104.02 (4)^\circ$ ,  $\gamma = 108.20 (4)^\circ$ ,  $Z = 2$  at 130 K, least-squares refinement of 228 parameters with 3951

reflections,  $R = 0.084$ . The complexes share a basic five-coordinate structure with approximately square-pyramidal geometry

and an apical chloride. The thiophene ring is bent from the plane of the remainder of the STPP" ligand core and is  $\eta^1$ -bound

to the metal through a pyramidal sulfur.

Adres publiczny

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<https://www.acs.org/content/acs/en.html>

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