

Redox-triggered helicity inversion in chiral cobalt complexes in combination with H⁺ and NO₃⁻ stimuli.

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Three chiral ligands with variable denticity, H₂L₂-H₂L₄, conjugated by *N,N'*-ethylenebis[*N*-methyl-(*S*)-alanine] and an ortho-heterosubstituted aromatic amine, were newly synthesized as analogues of previously reported H₂L₁. Four contracted-Λ_{oxo} cobalt(III) complexes [Co(L)]⁺ with left-handed helical structure of Λ₄Δ₂ configuration were prepared by one-electron oxidation of the corresponding contracted-Λ_{red} cobalt(II) complexes [Co(L)], which were generated from chiral ligands and Co(ClO₄)₂·6H₂O or Co(CF₃SO₃)₂·5.2H₂O in the presence of an organic base. Although the prepared cobalt(III) complexes were very inert and kinetically stable against protonation and NO₃⁻ complexation, cobalt(III) reduction in the presence of CF₃SO₃H and/or Bu₄NNO₃ allowed immediate changing of their three-dimensional structures from the contracted-Λ_{oxo} form to the extended-Λ [Co(H₂L)Y₂]^{*n*+} (Y = solvent and/or anion, *n* = 0-2) form with left-handed helicity or to the extended-Δ [Co(H₂L)(NO₃)]⁺ form with right-handed helicity via N- to O-amide coordination switching. Both extended forms were contracted to the original Λ_{oxo} form by oxidation of the cobalt(II) center in the presence of an organic base. Thus, redox reactions triggered dynamic helicity inversion of the chiral cobalt complexes, via multiple molecular motions consisting of relaxation/compression, extension/contraction, and helicity inversion motions in combination with deprotonation/protonation of amide linkages and NO₃⁻ anion complexation.

Adres publiczny

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<https://www.acs.org/content/acs/en.html>

