

Synthesis and characterization of $[\text{Me}_2\text{NC}(\text{S})\text{NP}(\text{S})(\text{OiPr})_2]^-$ complexes of Co(II), Ni(II), Zn(II) and Cd(II).

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Streszczenie

Reaction of the potassium salt of *N*-thiophosphorylthiourea $\text{Me}_2\text{NC}(\text{S})\text{NHP}(\text{S})(\text{OiPr})_2$ (**HL**) with Co(II), Ni(II), Zn(II) and Cd(II) cations in aqueous EtOH leads to $\text{M}(\text{L}-\text{S},\text{S}')_2$ (**ML₂**) chelate complexes. The structures of the resulting compounds were studied by IR, UV–Vis, ^1H , $^{31}\text{P}\{^1\text{H}\}$ NMR spectroscopy and microanalysis. The metal center is found in a tetrahedral environment in complexes **CoL₂**, **ZnL₂**, **CdL₂** and square-planar in **NiL₂** formed by the CS and PS sulfur atoms of two deprotonated ligands **L**. According to the NMR and UV–Vis spectroscopy data it was established that the metal cation of **NiL₂** is in a square-planar environment in CH_2Cl_2 , CHCl_3 and C_6H_6 , whereas **NiL₂** shows features of tetrahedral or octahedral complexes in acetone, DMSO and DMF. Molecular structures of complexes were elucidated by single crystal X-ray diffraction analysis.

Słowa kluczowe

cobalt, nickel, zinc, cadmium, *N*-thiophosphorylthiourea, crystal structures

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