

A short intramolecular hydrogen bond is a key factor in the self-assembly of a dimeric complex with a 22-membered metallamacrocyclic cavity.

Autorzy

Igor O. Fritsky

Henryk Kozłowski

Elena V. Prisyazhnaya

Aldona Karaczyn

Valentina A. Kalibabchuk

Tadeusz Głowiak

Rok wydania

1998

CzasopismoJournal of the Chemical
Society, Dalton TransactionsStrony

1535-1536

DOI

10.1039/A801781F

Kolekcja

Naukowa

Język

Angielski

Typ publikacji

Artykuł

Streszczenie

The reaction of copper(II) perchlorate with the new open chain oxime–amide ligand, *N,N'*-bis(2-hydroxyiminopropionyl)butane-1,4-diamine (H₂pab) in aqueous solution produced the dimeric complex [Cu(Hpab)]₂[ClO₄]₂·4H₂O containing a 22-membered macrocyclic cavity incorporating two copper(II) ions as a part of the ring framework.

Adres publiczny<https://doi.org/10.1039/A801781F>Strona internetowa wydawcy<https://www.rsc.org/>