

A dinuclear zinc(II) complex of a new unsymmetric ligand with an N₅O₂ donor set; a structural and functional model for the active site of zinc phosphoesterases.

Autorzy

Biswanath Das

Henrik Daver

Monika Pyrkosz-Bulska

Elke Persch

Suman K. Barman

Rabindranath Mukherjee

Elżbieta Gumienna-Kontecka

Martin Jarenmark

Fahmi Himu

Ebbe Nordlander

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Streszczenie

The dinuclear complex [Zn₂(DPCPMP)(pivalate)](ClO₄), where DPCPMP is the new unsymmetrical ligand [2-(N-(3-((bis((pyridin-2-yl)methyl)amino)methyl)-2-hydroxy-5-methylbenzyl)-N-((pyridin-2-yl)methyl)amino)acetic acid], has been synthesized and characterized. The complex is a functional model for zinc phosphoesterases with dinuclear active sites. The hydrolytic efficacy of the complex has been investigated using bis-(2,4-dinitrophenyl)phosphate (BDNPP), a DNA analog, as substrate. Speciation studies using potentiometric titrations have been performed for both the ligand and the corresponding dizinc complex to elucidate the formation of the active hydrolysis catalyst; it reveals that the dinuclear zinc(II) complexes, [Zn₂(DPCPMP)](2+) and [Zn₂(DPCPMP)(OH)](+) predominate the solution above pH4. The relatively high pK_a of 8.38 for water deprotonation suggests that a terminal hydroxide complex is formed. Kinetic investigations of BDNPP hydrolysis over the pH range 5.5-11.0 and with varying metal to ligand ratio (metal salt:ligand=0.5:1 to 3:1) have been performed. Variable temperature studies gave the activation parameters ΔH(‡)=95.59kJmol⁻¹, ΔS(‡)=-44.82Jmol⁻¹K⁻¹, and ΔG(‡)=108.00kJmol⁻¹. The cumulative results indicate the hydroxido-bridged dinuclear Zn(II) complex [Zn₂(DPCPMP)(μ-OH)](+) as the effective catalyst. The mechanism of hydrolysis has been probed by computational modeling using density functional theory (DFT). Calculations show that the reaction goes through one concerted step (S_N2 type) in which the bridging hydroxide in the transition state becomes terminal and performs a nucleophilic attack on the BDNPP phosphorus; leaving group dissociates simultaneously in an overall inner sphere type activation. Calculated free energy barrier is in good agreement with the experimentally determined activation parameters.

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