



Synthesis, crystal structure and magnetic properties of cyanide bridged 2D coordination polymers $[\text{Mn}(\text{salen})]_2[\text{Fe}(\text{CN})_5\text{NO}]$ and $[\text{Mn}(\text{salen})]_2[\text{Ni}(\text{CN})_4]$.

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The synthesis, crystal structure and magnetic properties of cyanide bridged 2D coordination complexes $[\text{Mn}(\text{salen})]_2[\text{Fe}(\text{CN})_5\text{NO}]$ 1 and $[\text{Mn}(\text{salen})]_2[\text{Ni}(\text{CN})_4]$ 2 are reported. Complex 1 crystallizes in the orthorhombic space group $Ccca$ ($a = 20.715$ (4) Å, $b = 20.718$ (4) Å, $c = 17.131$ (3) Å, $\alpha = \beta = \gamma = 90^\circ$, $Z = 8$). Its structure displays an extended 2D neutral network formed by cyclic tetrameric $[-\text{Mn}-\text{NC}-\text{Fe}-\text{CN}-]_4$ units. Complex 2 also crystallizes in the orthorhombic space group $Ccca$ ($a = 20.390$ (4) Å, $b = 20.701$ (4) Å, $c = 16.779$ (3) Å, $\alpha = \beta = \gamma = 90^\circ$, $Z = 8$) and shows an extended 2D neutral network formed by cyclic tetrameric $[-\text{Mn}-\text{NC}-\text{Ni}-\text{CN}-]_4$ units. The variable temperature magnetic susceptibility studies for both the complexes have been carried out in the temperature range 2-300 K which shows very weak antiferromagnetic interaction mediated by diamagnetic $[\text{Fe}(\text{CN})_5\text{NO}]^{2-}$ and $[\text{Ni}(\text{CN})_4]^{2-}$ ions.

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