

Deuterium isotope effect on ^{13}C NMR spectra of ortho Mannich bases.

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Rok wydania

1995

Czasopismo

Journal of Molecular Liquids

Numer woluminu

67

Strony

63-69

DOI

10.1016/0167-
7322(95)00866-7

Kolekcja

Naukowa

Język

Angielski

Typ publikacji

Artykuł

Streszczenie

The deuterium isotope effect upon ^{13}C NMR spectra was studied for ortho Mannich bases of varying ΔpK_a value [$\Delta\text{pK}_a = \text{pK}_a(\text{NH}^+) - \text{pK}_a(\text{OH})$]. The Δ values [$\Delta = \delta^{13}\text{C}(\text{OH}) - \delta^{13}\text{C}(\text{OD})$] for different positions in the phenyl ring are consistent with calculations of charge distribution by means of MNDO — PM3 method and assuming that deuteration leads to a weakening of the hydrogen bonding. In the critical range of ΔpK_a , where the proton transfer appears, the inversion of the sign of the isotope effect for the carbon atom in position 1 takes place. Strong dependence of the Δ value on temperature was revealed for 2-(N,N-diethylaminomethyl)-3,4,5,6-tetrachlorophenol showing the proton transfer equilibrium.

Adres publiczny

[https://doi.org/10.1016/0167-7322\(95\)00866-7](https://doi.org/10.1016/0167-7322(95)00866-7)

Strona internetowa wydawcy

<http://www.elsevier.com>